## Transformations of Penicillins and Cephalosporins: Reactions of Chlorosulphonyl Isocyanate with Penams, Cephems, and Azetidin-2-ones

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Benzyl 6 6 -phthalimidopenicillanate reacted with chlorosulphonyl isocyanate to give a thiazolo [3.2-c]pyrimidine derivative by ring expansion of the $\beta$-lactam. The trimethylsilyl derivative of $2.2,2$-trichloroethyl $6 \beta$-phenoxyacetamidopenicillanate gave two isomeric thiazolo[3.2-c]pyrimidine derivatives. Methyl 3 -methyl- $7 \beta$-phenoxy-acetamidoceph-3-em carboxylate gave a pyrimido [6.1-b][1.3]thiazine derivative. ( $3 R, 4 R$ )-3-Phthalimido1 -(1-benzyloxycarbonyl-2-methylprop-1-enyl)-4-methylthioazetidin-2-one did not give a corresponding $\beta$-lactam ring expansion product. but gave epimeric 4 -chloro-products by replacement of the 4 -methylthio-group. 4-Acetoxyazetidin-2-one reacted with chlorosulphonyl isocyanate to give an $N$-allophanoyl derivative.

Chlorosulphonyl isocyanate ${ }^{1}$ (CSI) reacts with an extensive series of organic functional groups, effecting a wide range of transformations. It undergoes nucleophilic additions at either the chlorosulphonyl or the isocyanate function with, for example, alcohols, amines, thiols, carboxylic acids and will also react with electron-
rich aromatic and heteroaromatic compounds. Enolizable ketones may be converted into $\beta$-ketonitriles, ${ }^{2}$ and

[^0]also into oxathiazinone, 1,3 -oxazine, and uracil derivatives. ${ }^{3}$ CSI also reacts with olefins to give azetidin-2ones, ${ }^{4}$ with imines to give triazindiones, ${ }^{1}$ and with
conversion of cephamycin derivatives ( 1 ) into the carbamates (2), ${ }^{6}$ and the transformation of the penicillanic acid (3) into the anhydropenicillin (4). ${ }^{7}$ A patent ${ }^{5}$ describes

(1) $R^{2}=H$
(2) $R^{2}=\mathrm{CONH}_{2}$

(7)

(10) $\mathrm{R}^{1}=\mathrm{PhO} \cdot \mathrm{CH}_{2} \cdot \mathrm{CO}, \mathrm{R}^{2}=\mathrm{CH}_{2} \cdot \mathrm{CCl}_{3}$

(13) $\mathrm{R}^{1}=$ acyl. $\mathrm{R}^{2}=\mathrm{NCO}$

(16) $\mathrm{R}^{1}=$ phthalimido, $\mathrm{R}^{2}=\mathrm{PhCH}_{2}$

(5) $x=0$
(6) $X=\mathrm{NSO}_{2} \mathrm{Cl}$

(11)
(17) $\mathrm{R}^{1}=$ phthalimido, $\mathrm{R}^{2}=\mathrm{PhCH}_{2}$


(15)

(20)

(9)

(12)



(4) $\mathrm{Rl}^{1}=$ trityl



(14) $\mathrm{R}^{1}=\mathrm{PhO} \cdot \mathrm{CH}_{2} \cdot \mathrm{CO}, \mathrm{R}^{2}=\mathrm{Me}$
(19a,b) $\mathrm{R}^{1}=$ trityl, $\mathrm{R}^{2}=\mathrm{PhCH}_{2}$

(21)
(18) $\mathrm{R}^{1}=$ trityl, $\mathrm{R}^{2}=\mathrm{PhCH}_{2}$
certain $N N$-disubstituted amides with replacement of the carbonyl group by the chlorosulphonylimino-group by a $[2+2]$ cycloaddition-cycloreversion sequence. ${ }^{5}$

Specific uses of CSI in $\beta$-lactam chemistry include the

[^1]the conversion of azetidin-2-ones (5) into the chloro-sulphonylimino-analogues (6). We now describe a new group of reactions of CSI with penams, cephems, and azetidin-2-ones.

Methyl 6及-phthalimidopenicillanate (7) reacted readily with excess of CSI at room temperature in methylene

[^2]chloride-ether. Reductive work-up with aqueous sodium metabisulphite (to remove the chlorosulphonyl group) and chromatography gave (8) (37\%). In the ${ }^{1} \mathrm{H}$ n.m.r. spectrum coupling between $\mathrm{H}-8$ and $-9(6 \mathrm{~Hz})$ could be compared with the cis-coupling ( 4 Hz ) between $\mathrm{H}-5$ and -6 in (7). The ${ }^{13} \mathrm{C}$ n.m.r. spectrum was also
patented reaction ${ }^{10}$ of penicillanate $S$-oxides with acyl isocyanates to give the 8,9-didehydro-analogues (13) which are schistosomicidal agents. De-esterification of of (11) and (12) gave the carboxylic acids which were not significantly active as antibiotics.

The $7 \beta$-phenoxyacetamidocephem ester (14) was

Table 1

| $\left[\mathrm{Eu}(\mathrm{fod})_{3}\right]:$ [substrate] |  |  |  |  | Gradient | Intercept <br> ( $\delta$ ) | Correlation coefficient |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Compound (11) | 0.049 | $\begin{array}{r} 0.078 \\ \text { Chem } \end{array}$ | $\begin{array}{r} 0.127 \\ \text { hift }(\delta) \end{array}$ | 0.144 |  |  |  |
| $2 \mathrm{a}-\mathrm{Me}$ | 1.66 | 1.65 | 1.66 | 1.66 | 0.36 | 1.61 | 0.72 |
| $2 \mathrm{~b}-\mathrm{Me}$ | 1.51 | 1.52 | 1.54 | 1.56 | 0.63 | 1.47 | 0.98 |
| $3-\mathrm{H}$ | 4.74 | 4.81 | 4.89 | 4.94 | 2.28 | 4.61 | 0.99 |
| 6-H | 9.95 | 10.18 | 10.46 | 10.54 | 7.79 | 9.41 | 0.96 |
| $8-\mathrm{H}$ | 5.14 |  |  | 5.84 | 6.77 | 4.85 |  |
| $9-\mathrm{H}$ | 5.49 | 5.50 | 5.52 | 5.56 | 0.43 | 5.48 | 0.86 |

in accord with structure (8), the presence of five carbonyl groups eliminating certain other possibilities. The structural similarity of (8) to thalidomide (9) is of interest, and this reaction type may therefore allow access to a range of analogues of central nervous system depressant drugs.
observed to react with CSI (without prior silylation), but refluxing di-isopropyl ether was needed, whereas the penicillanates reacted at room temperature. The elemental composition indicated incorporation of an HNCO moiety, and spectroscopic analysis suggested the pyrimido $[6,1-b]$ thiazine structure (15). Only one isomer

Table 2

|  | $\left[\mathrm{Eu}(\mathrm{fod})_{3}\right]:[$ substrate $]$ |  |  |  |  |  |  |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Compound <br> $(12)$ | 0.03 | 0.06 <br> Chemical | 0.09 <br> shift $(\delta)$ | 0.13 |  | Gradient | Intercept <br> $(\delta)$ |
| Correlation <br> coefficient |  |  |  |  |  |  |  |
| $2 \mathrm{a}-\mathrm{Me}$ | 1.72 | 1.74 | 1.77 | 1.80 | 0.92 | 1.68 | 0.99 |
| $3-\mathrm{H}$ | 1.56 | 1.61 | 1.66 | 1.70 | 1.52 | 1.52 | 0.99 |
| $6-\mathrm{H}$ | 4.97 | 5.11 | 5.30 | 5.44 | 5.24 | 4.80 | 0.99 |
| $8-\mathrm{H}$ | 8.97 | 9.04 | 9.14 | 9.24 | 3.21 | 8.84 | 0.99 |
| $9-\mathrm{H}$ | 5.33 | 5.45 | 5.70 | 5.78 | 7.33 | 4.98 | 0.99 |
|  | 5.62 | 5.78 | 5.97 | 6.09 | 5.11 | 5.46 | 0.99 |

In a similar reaction, trichloroethyl $6 \beta$-phenoxyacetamidopenicillanate (10) was first trimethylsilylated to protect the $6 \beta$-amide (probably as the imidate trimethylsilyl derivative). Reaction in situ with CSI, treatment with metabisulphite, and chromatography gave two major non-polar reaction products, shown to be the $\mathrm{C}-8$ phenoxyacetamido-epimers (11) and (12). The ${ }^{1} \mathrm{H}$ coupling constants between H-8 and -9 in (ll) ( 4 Hz ) suggested cis-orientation, whereas the corresponding coupling in (12) ( 11 Hz ) was in accord with transorientation. Lanthanide shift studies (Tables 1 and 2) confirmed these stereochemical assignments.* Consideration of the relative rates of shift of $\mathrm{H}-8$ and -9 in each isomer, as a function of the molar ratio of shift reagent, showed that $\mathrm{H}-9$ is trans to the phenoxyacetamidogroup in the less polar isomer (11), and cis in the more polar isomer (12). The isomerism at C-8 in (11) and (12) probably occurred during the silylation process. ${ }^{8}$ The mechanism of the $\beta$-lactam ring expansion is unclear, but precedent exists in the reaction of thiocyanic acid with penicillins to give a related 5 -thiouracil, ${ }^{9}$ and in a

[^3]was formed, in contrast to the reaction of (10), H-9 and -10 being tentatively assigned cis-stereochemistry.

The reaction of CSI with monocyclic azetidinones such as (16) was of interest because of the possibility of formation of 2 -imino analogues ${ }^{5}$ by a $[2+2]$ cyclo-addition-cycloreversion sequence, and also because of possible ring-expansion to give substituted uracils. Additionally, reaction at the 4 -methylthio-group was envisaged. $3 \beta$-Phthalimidosecopenicillanate (16) was therefore refluxed in di-isopropyl ether with CSI, giving almost quantitatively a mixture of 4 -chloroazetidin-2ones (17), with cis: trans ratio of $2: 3\left(J_{3.4 \text { cis }} 4, J_{3,4 \text { trans }}\right.$ 1.2 Hz ). The mechanism could involve a 'Type 3 ' 1 reaction of the chlorosulphonyl group (Scheme). Other mechanisms involving initial attack by the methylthiogroup on the isocyanate moiety of CSI can, however, be envisaged. It was not possible to separate the isomeric 4 -chloroazetidin-2-ones (17), and the absolute stereochemistries at $\mathrm{C}-3$ and -4 were therefore not assigned. The $3 \beta$-tritylaminoazetidin-2-one (18) when subjected to a similar CSI reaction did not give the corresponding 4-

[^4]chloroazetidin-2-ones, but gave an inseparable mixture.*

4-Acetoxyazetidin-2-one (20) has recently received much attention in the synthesis of analogues of the $\beta$ lactam antibiotics. ${ }^{3,11}$ When (19) was treated in ether with CSI and the resultant product treated with aqueous sodium metabisulphite (to hydrolyse chlorosulphonyl groups), a $\beta$-lactam product of elemental composition $\mathrm{C}_{7} \mathrm{H}_{9} \mathrm{~N}_{3} \mathrm{O}_{5}$ was obtained, suggesting incorporation of two isocyanate units. The ${ }^{1} \mathrm{H}$ n.m.r. spectrum was
give by a third reaction mode nitrogen adducts which may be of use in the synthesis of analogues of the $\beta$ lactam antibiotics.

## EXPERIMENTAL

General details are as reported previously.
Reaction of Methyl 63-Phthalimidopenicillanate (7) with Chlorosulphonyl Isocyanate.-Methyl phthalimidopenicillanate ( $2 \mathrm{~g}, 5.55 \mathrm{mmol}$ ) in dry dichloromethane was added to chlorosulphonyl isocyanate ( $0.83 \mathrm{~g}, 0.5 \mathrm{ml}, 6 \mathrm{mmol}$ ) in


Scheme
characteristic of a 4-acetoxyazetidin-2-one, and, together with the ${ }^{13} \mathrm{C}$ n.m.r. spectrum indicated structure (21) rather than alternative uracil-based structures. The mass spectral fragmentation modes were also in accord with this structure (Figure). A related reaction

which has been reported ${ }^{12}$ involved the addition of a single chlorosulphonyl isocyanate unit to the NH group of an azetidin-2-one.

Reactions of chlorosulphonyl isocyanate with penicillanates and cephalosphoranates therefore lead to two relatively unusual fused uracil ring systems, thiazolo-$[3,2-c]$ pyrimidines and pyrimido[6,1-b][1,3]thiazines. 4-Methylthioazetidin-2-ones, however, do not undergo $\beta$ lactam ring expansion under similar reaction conditions, but give instead 4 -chloroazetidin- 2 -ones by a different reaction pathway. l-Unsubstituted azetidin-2-ones

* The $3 \beta$-tritylaminoazetidin-2-one could, however, be converted into a separable mixture ( $5: 1$ ) of cis- and trans-4-chloro-azetidin-2-ones (19) in good yield by the reaction of (dichloroiodo) benzene (see Experimental section).
dry diethyl ether ( 1 ml ) with stirring at room temperature. After 1.5 h a further portion of chlorosulphonyl isocyanate ( 6 mmol ) was added, and after 3 h the mixture was diluted with chloroform ( 50 ml ) and aqueous sodium metabisulphite added. The aqueous layer was extracted with further portions of chloroform and the combined chloroform extracts dried $\left(\mathrm{MgSO}_{4}\right)$, evaporated in vacuo, and chromatographed to give, as a solid which melted with decomposition over a wide temperature range, ( $3 \mathrm{R}, 8 \mathrm{R}, 8 \mathrm{aR}$ )methyl 2,2-dimethyl-5,7-dioxo-8-phthalimidoperhydrothiazolo-$\left[3,2\right.$-c]pyrimidine-3-carboxylate (8) ( $0.85 \mathrm{~g}, 37 \%$ ), $[\alpha]_{\mathrm{D}}{ }^{20}$ $+15^{\circ}\left(c 1.2 \mathrm{CHCl}_{3}\right), \nu_{\text {max. }}(\mathrm{KBr}) 3350(\mathrm{NH}), 1780$ and 1725 (phthalimido $\mathrm{C}=\mathrm{O}$ ), and $1715 \mathrm{~cm}^{-1}$ (amide $=\mathrm{O}$ ), $\delta_{\mathrm{H}}$ $\left(\mathrm{CDCl}_{3}\right) 1.38$ and 1.33 (each $\left.3 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{2}\right), 3.80(3 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{CO}_{2} \mathrm{Me}\right), 4.62(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H}), 5.02(1 \mathrm{H}, \mathrm{d}, J 6 \mathrm{~Hz}, 8-\mathrm{H})$, $6.87(1 \mathrm{H}, \mathrm{d}, J 6 \mathrm{~Hz}, 9-\mathrm{H}), 7.85(4 \mathrm{H}, \mathrm{m}$, phthalimide), and $8.37 \mathrm{br}(1 \mathrm{H}, \mathrm{s}, \mathrm{NH}), \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 169.18,167.48,166.94$, 165.28 and 150.48 (each, $\mathrm{C}=\mathrm{O}$ ), 134.68 (d), 131.34 (s), 131.26 (s), 124.09 (d), and 123.81 (d) (phthalimido carbons), 70.08 (d, C-8), 63.20 (d, C-9), 52.89 (s, C-2), 52.45 (q, $\mathrm{CO}_{2} \mathrm{Me}$ ), 50.75 (d, C-3), and 33.22 and 26.03 p.p.m. (each q, $\mathrm{CMe}{ }_{2}$ ) (Found: $M^{+}, 403.0838 . \mathrm{C}_{18} \mathrm{H}_{17} \mathrm{~N}_{3} \mathrm{O}_{6} \mathrm{~S}$ requires $M$, 403.0838 ), $m / e 403$ ( $M^{+}, 8 \%$ ), 344 (15), 289 (35), 256 (100), 174 (65), and 116 (60).
Reaction of 2,2,2-Trichloroethyl 6 $\beta$-Phenoxyacetamidopenicillanate (10) with Chlorosulphonyl Isocyanate.-NOBis(trimethylsilyl)acetamide ( $2.5 \mathrm{~g}, 12.3 \mathrm{mmol}$ ) was added
${ }^{11}$ A. K. Mukerjee and R. C. Srivasta, Synthesis, 1973, 327; H. W. Schnabel, D. Grimm, and H. Jensen, Annalen, 1974, 477; D. Bormann, ibid., 1974, 1391; A. G. Brown, D. F. Corbett, and T. T. Howarth, J.C.S. Chem. Comm., 1977, 359.

12 R. Graf, ref. 1, p. 178.
to ( 10 ) ( $6 \mathrm{~g}, 12.2 \mathrm{mmol}$ ) in dry dichloromethane ( 50 ml ) and stirred for 2 h at room temperature under nitrogen. Chlorosulphonyl isocyanate ( $3.57 \mathrm{~g}, 2.2 \mathrm{ml}, 25.3 \mathrm{mmol}$ ) was added to the solution at $0^{\circ}$, stirred for 5 min , and the mixture allowed to come to room temperature and stirred for a further 1.5 h . The solution was shaken with concentrated aqueous sodium metabisulphite and the dichloromethane solution dried $\left(\mathrm{MgSO}_{4}\right)$. Column chromatography yielded a mixture of two isomeric compounds, followed by a second pair of products of greater polarity.

The first pair of products was rechromatographed to yield first ( $3 \mathrm{R}, 8 \mathrm{R}, 8 \mathrm{aR}$ )-2,2,2-trichloroethyl 2,2-dimethyl-5,7-dioxo-8-phenoxyacetamidoperhydrothiazolo $[3,2$-c]pyrimi-dine-3-carboxylate (11) ( $1.1 \mathrm{~g}, 17 \%$ ), m.p. $113.5-114^{\circ}$, $[\alpha]_{\mathrm{D}}{ }^{22}+2.6^{\circ}\left(c 1.17 \mathrm{CHCl}_{3}\right), \nu_{\text {max. }}(\mathrm{KBr}) 3300 \mathrm{br}$ and 3200 br (amide NH), 1750 (ester $\mathrm{C}=\mathrm{O}$ ), 1720 and 1700 (amide $\mathrm{C}=\mathrm{O}$ ), and $1430 \mathrm{~cm}^{-1}, \delta_{\mathrm{H}}\left[\mathrm{C}_{6} \mathrm{D}_{6}-\mathrm{CD}_{3} \mathrm{COCD}_{3}(1: 1)\right] 1.72$ and 1.59 (each $3 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{2}$ ), $4.73\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{CCl}_{3}\right), 4.82$ $\left.(2 \mathrm{H}, \mathrm{s}, \mathrm{PhOCH})_{2}\right), 4.97(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H}), 5.32(1 \mathrm{H}, \mathrm{dd}, J 9.5$ and $4 \mathrm{~Hz}, 8-\mathrm{H}), 5.83(1 \mathrm{H}, \mathrm{d}, J 4 \mathrm{~Hz}, 9-\mathrm{H}), 7.25(5 \mathrm{H}, \mathrm{m}$, $\mathrm{Ph}), 8.59(1 \mathrm{H}, \mathrm{d}, J 9.5 \mathrm{~Hz}$, side chaim amide), and 9.92 ( $1 \mathrm{H}, \mathrm{s}$, ring NH) (Found: $M^{+}, 523.0152 . \mathrm{C}_{19} \mathrm{H}_{20}{ }^{35} \mathrm{Cl}_{3}{ }^{-}$ $\mathrm{N}_{3} \mathrm{O}_{6} \mathrm{~S}$ requires $M, 523.0139$ ), m/e $523\left(M^{+}, 2\right), 489$ (7), 449 (9), 372 (18), 348 (12), 290 (20), 234 (13), 185 (55), and 151 ( $100 \%$ ) (Found: C, 43.7 ; H, 3.7; N, 8.0. $\mathrm{C}_{19} \mathrm{H}_{20} \mathrm{Cl}_{3}{ }^{-}$ $\mathrm{N}_{3} \mathrm{O}_{6} \mathrm{~S}$ requires $\mathrm{C}, 43.5 ; \mathrm{H}, 3.8 ; \mathrm{N}, 8.0 \%$ ). The isomer (12) $(0.45 \mathrm{~g}, 17 \%),(3 \mathrm{R}, 8 \mathrm{~S}, 8 \mathrm{aR})-2,2,2$-trichlovoethyl $2,2-$ dimethyl-5,7-dioxo-8-phenoxyacetamidoperhydrothiazolo-
[3,2-c]pyrimidine-3-carboxylate, was then obtained as an amorphous solid, $[\alpha]_{\mathrm{D}}{ }^{22}+23^{\circ}\left(c \quad 1.075 \mathrm{CHCl}_{3}\right)$, $\nu_{\text {max. }}(\mathrm{KBr})$ 3400-3200 (unresolved NH), 1750 (ester $\mathrm{C}=\mathrm{O}$ ), 1710 (unresolved amide $\mathrm{C}=\mathrm{O}$ ), and $1440 \mathrm{~cm}^{-1}, \delta_{\mathrm{H}}\left[\mathrm{C}_{6} \mathrm{D}_{6}-\mathrm{CD}_{3}-\right.$ $\left.\mathrm{COCD}_{3}(1: 1)\right] 1.58$ and 1.42 (each $3 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{2}$ ), $4.46(2 \mathrm{H}$, $\left.\mathrm{s}, \mathrm{CH}_{2} \mathrm{CCl}_{3}\right), 4.75(2 \mathrm{H}, \mathrm{s}, \mathrm{PhOCH} 2), 4.89(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H})$, $5.10(1 \mathrm{H}, \mathrm{dd}, J 9$ and $11 \mathrm{~Hz}, 8-\mathrm{H}), 5.80(1 \mathrm{H}, \mathrm{d}, J 11 \mathrm{~Hz}$, $9-\mathrm{H}), 7.4-6.7(5 \mathrm{H}, \mathrm{m}, \mathrm{PhO}), 8.08(1 \mathrm{H}, \mathrm{d}, J 9 \mathrm{~Hz}$, side chain NH ), and $9.67\left(1 \mathrm{H}, \mathrm{s}\right.$, ring NH ) (Found: $M^{+}$, 523.0113 . $\mathrm{C}_{19} \mathrm{H}_{20}{ }^{35} \mathrm{Cl}_{3} \mathrm{~N}_{3} \mathrm{O}_{6} \mathrm{~S}$ requires $M$, 523.0139 ), $m / e$ 523 (2\%), 449 (3), 372 (6), 348 (5), 290 (8), 262 (4), 261 (8), 198 (18), and $151(100 \%)$. Mixed fractions containing (11) and (12) ( 0.55 g ) were also obtained.

100 MHz N.m.r. spectra were measured for (11) and (12) with varying molar ratios of $\mathrm{Eu}(\mathrm{fod})_{3}$ to substrate. The chemical shift of each proton was plotted against the molar ratio of $\mathrm{Eu}(\mathrm{fod})_{3}$ and the gradient of each plot calculated, allowing assignment of stereochemistry.

The second isomeric pair of products were not separated by chromatography, but were obtained as a mixture $(0.4 \mathrm{~g}$, $6 \%$ ), $\nu_{\text {max. }}(\mathrm{KBr}) 1800 \mathrm{~cm}^{-1}$ (Found $M^{+}, 523.0130$. Calc. for $\mathrm{C}_{19} \mathrm{H}_{20}{ }^{35} \mathrm{Cl}_{3} \mathrm{~N}_{3} \mathrm{O}_{6} \mathrm{~S}: M, 523.013$ 9.)

De-esterification of (11) and (12).-Ester (11) (0.17 g, 0.32 mmol ) in dimethylformamide-glacial acetic acid ( $25: 7.5$ ) ( 11 ml ) was stirred for 2 h at $0^{\circ}$ with zinc dust $(0.53 \mathrm{~g}, 8.1 \mathrm{mmol})$. The mixture was filtered, diluted with ethyl acetate, and washed with water. The organic layer was dried $\left(\mathrm{MgSO}_{4}\right)$, evaporated, and the residue crystallized from chloroform to give the carboxylic acid, m.p. $125^{\circ}$ (decomp.), $[\alpha]_{\mathrm{D}}{ }^{22}+6.8^{\circ}\left(c 1.03 \mathrm{CHCl}_{3}\right), v_{\max .}(\mathrm{KBr}) 3600-$ $2700\left(\mathrm{CO}_{2} \mathrm{H}\right), 3300$ and $3200(\mathrm{NH})$, and $1750-1650 \mathrm{br}$ $(\mathrm{C}=\mathrm{O}), \delta\left(\mathrm{CDCl}_{3}\right) 1.67$ and 1.55 (each $3 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{2}$ ), 4.70 $(2 \mathrm{H}, \mathrm{s}), 4.83(1 \mathrm{H}, \mathrm{s}), 5.1(1 \mathrm{H}, \mathrm{dd}), 5.84(1 \mathrm{H}, \mathrm{d}), 7.7-$ $6.9(5 \mathrm{H}, \mathrm{m}, \mathrm{PhO}), 8.35(1 \mathrm{H}, \mathrm{d}, J 9 \mathrm{~Hz}$, side chain NH$)$, and $9.58\left(1 \mathrm{H}, \mathrm{s}\right.$, ring NH), $m / e 393\left(M^{+}, 4 \%\right), 347$ (8), $319(10), 261$ (30), $160(40)$, and $132(30 \%)$.

Similarly prepared was the isomeric carboxylic acid,
m.p. $c a .125^{\circ}$ (decomp.), $[\alpha]_{\mathrm{D}}{ }^{22}+82^{\circ}\left(c 1.24 \mathrm{CHCl}_{3}\right), \nu_{\text {max. }}$ ( KBr ) $3600-2700 \mathrm{br}\left(\mathrm{CO}_{2} \mathrm{H}\right), 3350$ and $3200(\mathrm{NH})$, and 1730 and $1700 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}), \delta\left(\mathrm{CDCl}_{3}\right) 1.71$ and 1.51 (each $\left.\left.3 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{2}\right), 4.58(2 \mathrm{H}, \mathrm{s}, \mathrm{PhOCH})_{2}\right), 4.71(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H})$, $4.98(1 \mathrm{H}, \mathrm{dd}, J 9$ and 12 Hz$), 5.64(1 \mathrm{H}, \mathrm{d}, J 12 \mathrm{~Hz})$, $7.50-6.85(5 \mathrm{H}, \mathrm{m}, \mathrm{Ph}), 8.05(1 \mathrm{H}, \mathrm{d}, J 9 \mathrm{~Hz}$, side chain NH ), and $9.40\left(1 \mathrm{H}, \mathrm{s}\right.$, ring NH), $m / e 393\left(M^{+}, 3 \%\right), 375$ (7), 349 (6), 315 (70), 261 (60), and 243 ( $11 \%$ ).

Reaction of Methyl 3-Methyl-7 $\beta$-phenoxyacetamidoceph-3-em-4-carboxylate (14) with Chlorosulphonyl Isocyanate.Methyl ester (14) ( $0.2 \mathrm{~g}, 0.48 \mathrm{mmol}$ ) and NO-bis(trimethylsilyl)acetamide ( $0.19 \mathrm{~g}, 1.40 \mathrm{mmol}$ ) were stirred with chlorosulphonyl isocyanate ( $0.06 \mathrm{~g}, 0.4 \mathrm{mmol}$ ) at room temperature, but did not react after 2.5 h . Reflux led to a complex mixture.

The reaction was repeated, adding chlorosulphonyl isocyanate ( $0.114 \mathrm{~g}, 0.07 \mathrm{ml}, 0.8 \mathrm{mmol}$ ) dropwise to (14) $(0.122 \mathrm{~g}, 0.29 \mathrm{mmol})$ in dry di-isopropyl ether ( 25 ml ) under nitrogen. After refluxing for 3 h the solution was washed with aqueous sodium metabisulphite, extracted with chloroform ( $2 \times 50 \mathrm{ml}$ ) and ethyl acetate ( 50 ml ), and the combined extracts dried $\left(\mathrm{MgSO}_{4}\right)$, evaporated, and chromatographed. Following a second column chromatography (9R,9aR)-methyl 7,8,9,9a-tetrahydro-3-methyl-6,8-dioxo-10-phenoxyacetamido- $2 \mathrm{H}, 6 \mathrm{H}$-pyrimidino $[6,1-\mathrm{b}][1,3]$ -thiazine-4-carboxylate (15) ( $0.076 \mathrm{~g}, 56 \%$ ) was obtained as an oil, $\nu_{\max }$ (film) $3400,3240,1760,1740$, and $1590 \mathrm{~cm}^{-1}$, $\delta\left(\mathrm{CDCl}_{3}\right) 2.18(3 \mathrm{H}, \mathrm{s}, 3-\mathrm{Me}), 2.96(2 \mathrm{H}, \mathrm{ABq}, J 15 \mathrm{~Hz}$, $2-\mathrm{H}), 3.68(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 4.92(1 \mathrm{H}, \mathrm{d}, J 3 \mathrm{~Hz}, 10-\mathrm{H}), 5.08$ $\left(3 \mathrm{H}, \mathrm{m}, 1-\mathrm{H}\right.$ and $\left.\mathrm{PhOCH}_{2}\right), 6.5-5.9 \mathrm{br}(1 \mathrm{H}, \mathrm{s}, \mathrm{NH})$, 7.34-6.8 ( $6 \mathrm{H}, \mathrm{m}$ ), and 9.10br ( $1 \mathrm{H}, \mathrm{s}, \mathrm{NH}$ ) (Found: $M^{+}$, $405.0990 . \quad \mathrm{C}_{18} \mathrm{H}_{19} \mathrm{~N}_{3} \mathrm{O}_{6} \mathrm{~S}$ requires $\left.M^{+}, 405.0995\right), \quad \mathrm{m} / \mathrm{e}$ $405\left(M^{+}, 9 \%\right), 373(5), 371(5), 312(10), 261$ (8), 234 (28), $171(20)$, and 141 (60).

Reaction of (3R,4R)-3-Phthalimido-1-(1-benzyloxycarbonyl-2-methylprop-1-enyl)-4-methylthioazetidin-2-one (16) with Chlorosulphonyl Isocyanate.-Benzyl ester (16) (0.06 g, 0.13 mmol ) was refluxed with excess of chlorosulphonyl isocyanate in di-isopropyl ether-dichloromethane ( $1: 1$ ) and the reaction monitored by t.l.c. until after 8 h the starting material had been consumed. After washing the solution with aqueous sodium metabisulphite, drying $\left(\mathrm{MgSO}_{4}\right)$, evaporation, and repeated column chromatography the chloro-compounds (17) ( $0.058 \mathrm{~g}, 98 \%$ ) were obtained as an inseparable mixture and identified by mass spectrometry and spectroscopic analysis of the mixture.

Reaction of (18) with (Dichloroiodo)benzene.-The tritylaminosecopenicillanate (18) ( $0.76 \mathrm{~g}, 1.35 \mathrm{mmol}$ ) in dry pyridine ( 6 ml ) was cooled to $0^{\circ}$ and (dichloroiodo) benzene $(1.07 \mathrm{~g}, 4.05 \mathrm{mmol})$ added with stirring. After 15 min water ( 1 ml ) was added dropwise to the cooled solution $\left(-15^{\circ}\right)$ and after a further 15 min ethyl acetate ( 20 ml ) was added. The organic solution was washed four times with brine and with aqueous copper sulphate, dried $\left(\mathrm{MgSO}_{4}\right)$, and evaporated to give a brown oil (1.40 g). Column chromatography gave the trans-chloride (19a) ( 0.07 g ) as a solid, $[\alpha]_{\mathrm{D}}{ }^{22}+30^{\circ}$, m.p. $65^{\circ}, \nu_{\max }(\mathrm{KBr}) 3315,1780$, and $1730-1710 \mathrm{~cm}^{-1}, \delta\left(\mathrm{CDCl}_{3}\right) 8.30(20 \mathrm{H}, \mathrm{m}), 5.14(2 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{PhCH}_{2}\right), 4.76(1 \mathrm{H}, \mathrm{d}, J 1 \mathrm{~Hz}, 4-\mathrm{H}), 4.33 \mathrm{br}(1 \mathrm{H}, \mathrm{dd}, J 10$ and $1 \mathrm{~Hz}, 3-\mathrm{H}), 3.60(1 \mathrm{H}, \mathrm{d}, J 10 \mathrm{~Hz}, \mathrm{NH})$, and 2.28 and 1.92 (each $3 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{2}$ ) (Found: $M^{+}, 550.2022 . \mathrm{C}_{34} \mathrm{H}_{31^{-}}$ $\mathrm{ClN}_{2} \mathrm{O}_{3}$ requires $M^{+}, 550.202$ 7). The cis-chloride (19b) was obtained as a solid ( 0.40 g ), $[\alpha]_{\mathrm{D}}{ }^{22}+40^{\circ}$, m.p. $67^{\circ}$ (decomp.), $\nu_{\text {max. }}(\mathrm{KBr}) 1775$ and $1725 \mathrm{~cm}^{-1}, \delta\left(\mathrm{CDCl}_{3}\right) 7.20-7.60(2 \mathrm{H}$, $\mathrm{m}), 5.09\left(2 \mathrm{H}, \mathrm{s}, \mathrm{PhCH}_{2}\right), 4.45(1 \mathrm{H}, \mathrm{dd}, J 4$ and 10 Hz ,
$3-\mathrm{H}), 3.17(1 \mathrm{H}, \mathrm{d}, J 4 \mathrm{~Hz}, 4-\mathrm{H})$, and 2.25 and 1.98 (each $3 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{2}$ ) (Found: C, 73.9; H, 6.0; N, 4.9; Cl, 6.3. $\mathrm{C}_{34} \mathrm{H}_{31} \mathrm{ClN}_{2} \mathrm{O}_{3}$ requires $\mathrm{C}, 74.1 ; \mathrm{H}, 5.6 ; \mathrm{N}, 5.1 ; \mathrm{Cl}$, $6.5 \%$.)

Reaction of 4-Acetoxyazetidin-2-one (20) with Chlorosulphonyl Isocyanate.-The acetate ( 20 ) ( $0.25 \mathrm{~g}, 1.94 \mathrm{mmol}$ ) in dry ether ( 20 ml ) was purged with dry nitrogen and then chlorosulphonyl isocyanate ( $0.55 \mathrm{~g}, 0.34 \mathrm{ml}, 3.88 \mathrm{mmol}$ ) added dropwise at room temperature. After stirring at room temperature for 10 min , and then refluxing for 10 min , the solution was cooled and aqueous sodium metabisulphite ( $1 \%$ solution, 50 ml ) added dropwise. Vigorous reaction preceded precipitation of crystals which were filtered, washed with cold, dry ether, and dried to give (21) ( 0.13 g ), m.p. $157.5-160^{\circ}$, $\nu_{\text {max. }}(\mathrm{KBr}) 3400,3280,3180$,
$1800,1750,1730,1690$, and $\left.1610 \mathrm{~cm}^{-1}, \delta_{\mathrm{H}}\left({ }^{2}{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right)$ $7.3 \mathrm{br}, 6.4 \mathrm{br}$, and 5.3 br (each $1 \mathrm{H}, \mathrm{NH}$ ), $6.69(1 \mathrm{H}, \mathrm{dd}, J 2$ and $4 \mathrm{~Hz}, 4-\mathrm{H}), 3.41\left(1 \mathrm{H}, \mathrm{dd}, J 4\right.$ and $\left.16 \mathrm{~Hz}, 3-\mathrm{H}_{c i s}\right)$, and $2.96\left(1 \mathrm{H}, \mathrm{dd}, J 2\right.$ and $\left.16 \mathrm{~Hz}, 3-\mathrm{H}_{\text {trans }}\right), \delta_{\mathrm{C}}\left(\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right)$ 169.43, 164.61, 149.05, and 141.19 (each $\mathrm{C}=\mathrm{O}$ ), 73.58 (C-4), 40.64 (C-3), and 20.7 p.p.m. (C-6) \{cf. 4 -acetate (19), $\left.\delta_{\mathrm{C}}\left({ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right) 171.20(\mathrm{C}-2, \mathrm{~s}), 166.09$ (COMe, s), 73.13 (C-4, d), 44.98 (C-3, d), and 20.78 (COMe, q) \} (Found: C, 38.4; $\mathrm{H}, 4.4 ; \mathrm{N}, 19.3 . \mathrm{C}_{7} \mathrm{H}_{9} \mathrm{~N}_{3} \mathrm{O}_{5}$ requires $\mathrm{C}, 38.7 ; \mathrm{H}$, 4.2 ; $\mathrm{N}, 19.5 \%$ ), m/e $215\left(M^{+}\right)$.
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